DEPARTMENT OF CHEMISTRY

FEDERAL UNIVERSITY OF TECHNOLOGY, MINNA FIRST SEMESTER EXAMINATION 2012/2013 SESSION

COURSE CODE; CHM 315

COURSE TITLE: POLYMER CHEMISTRY

INSTRUCTIONS: ANSWER ANY FOUR QUESTION

TIME ALLOWED: 2 HOURS

3 UNITS

1. (a) Distingush between Block copolymer and Graft copolymer

(b). Complete the table below

Polymer	Monomer	Repeating unit	
Poly(Vinylchloride)			
Poly(Vinylacetate)		12	1000
Polystyrene			Action and a second
Polyisoprene			A COLOR DE LA COLO
Polytetrafluoroethylene			

- (c). Show by equation the polymerization of phenol and methanal to form a plenol-formaldehyde
- 2. (a). Differenciate between static and dynamic equlibrium
 - (b). State the Vant Hoff's equation and define all the terms
 - (c). Explain the operating principle of Osmometry
- 3. (a). Name and explain the working principle of Gel Permeation Chromatography.
 - (b). Enumerate any five draw back of Gel Permeation Chromatography.
 - (c). Explain briefly the folder chain lamellar model and extended crystalline model
- 4. (a). Define the mechanical properties of polymers
 - (b) Enumerate any three factors governing the mechanical properties of a particular polymer
- (c) List any four factors that induce polymer crystallinity

- 5. (a). Distinguish between Fibers and Plastics
- (b) What is the molecular weight of polyester if 1.2 cm3 of 0.1M of KOH were required for neutralization of 0.5 g of polymerization sample with COOH group as end group
 - (c) Write: short note on the following
 - (i) Industrial polymerization (ii) Tacticity